# **CETYL PALMITATE**

# Cetylis palmitas

### DEFINITION

Mixture of C<sub>14</sub>-C<sub>18</sub> esters of lauric (dodecanoic), myristic (tetradecanoic), palmitic (hexadecanoic) and stearic (octadecanoic) acids ('Cetyl esters wax').

*Content* (expressed as hexadecyl hexadecanoate): 10.0 per cent to 20.0 per cent for Cetyl palmitate 15, 60.0 per cent to 70.0 per cent for Cetyl palmitate 65 and minimum 90.0 per cent for Cetyl palmitate 95.

### **CHARACTERS**

Appearance: white or almost white, waxy plates, flakes or powder.

Solubility: practically insoluble in water, soluble in boiling anhydrous ethanol and in methylene chloride, slightly soluble in light petroleum, practically insoluble in anhydrous ethanol.

mp: about 45 °C for Cetyl palmitate 15 and Cetyl palmitate 65 and about 52 °C for Cetyl palmitate 95.

#### **IDENTIFICATION**

- A. It complies with the limits of the assay and the chromatogram obtained with the test solution shows the typical main peak(s).
- B. Saponification value (see Tests).

### TESTS

Appearance of solution. The solution is not more intensely coloured than reference solution  $Y_6$  (2.2.2, Method II).

Dissolve 4.0 g in *methylene chloride R* and dilute to 20 ml with the same solvent.

Acid value (2.5.1): maximum 4.0.

Dissolve 10.0 g in 50 ml of the solvent mixture described by heating under reflux on a water-bath for 5 min.

Hydroxyl value (2.5.3, Method A): maximum 20.0.

Iodine value (2.5.4, Method A): maximum 2.0.

Saponification value (2.5.6): 105 to 120.

Heat under reflux for 2 h.

Alkaline impurities. Dissolve 2.0 g 'with gentle heating' in a mixture of 1.5 ml of ethanol (96 per cent) R and 3 ml of toluene R. Add 0.05 ml of a 0.4 g/l solution of bromophenol blue R in ethanol (96 per cent) R. Not more than 0.4 ml of 0.01 *M hydrochloric acid* is required to change the colour of the solution to yellow.

Nickel (2.4.31): maximum 1 ppm.

Water (2.5.12): maximum 0.3 per cent, determined on 1.0 g using a mixture of equal volumes of anhydrous methanol R and methylene chloride R as solvent.

Total ash (2.4.16): maximum 0.2 per cent, determined on 1.0 g.

#### ASSAY

Gas chromatography (2.2.28): use the normalisation procedure.

*Test solution*. Dissolve 20.0 mg of the substance to be examined in *hexane R* and dilute to 20.0 ml with the same solvent.

01/2008:1906 Reference solution (a). Dissolve 20.0 mg of cetyl palmitate 95 CRS in hexane R and dilute to 20.0 ml with the same solvent.

> *Reference solution (b).* Dissolve 20.0 mg of cetyl palmitate 15 CRS in hexane R and dilute to 20.0 ml with the same solvent.

## Column:

- material: stainless steel;
- size: l = 10 m,  $\emptyset = 0.53 \text{ mm}$ ;
- stationary phase: poly(dimethyl)siloxane R (film thickness 2.65 µm).

Carrier gas: helium for chromatography R.

Flow rate: 6.5 ml/min.

Split ratio: 1:10.

#### Temperature:

	Time (min)	Temperature (°C)
Column	0 - 10	$100 \rightarrow 300$
	10 - 15	300
Injection port		350
Detector		350

Detection: flame ionisation.

Injection: 1 µl.

*Relative retention* with reference to cetyl palmitate (retention time = about 9 min): cetyl alcohol = about 0.3; palmitic acid = about 0.4; lauric ester = about 0.8; myristic ester = about 0.9; stearic ester = about 1.1.

*System suitability*: reference solution (b):

- *resolution*: minimum of 1.5 between the peaks due to cetyl palmitate and cetyl stearate.

#### STORAGE

At a temperature not exceeding 25 °C.

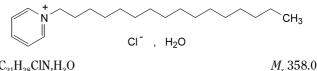
#### LABELLING

The label states the type of cetyl palmitate.

01/2008:0379 corrected 6.0

# **CETYLPYRIDINIUM CHLORIDE**

# Cetylpyridinii chloridum



C<sub>21</sub>H<sub>38</sub>ClN,H<sub>2</sub>O [6004-24-6]

### DEFINITION

Cetylpyridinium chloride contains not less than 96.0 per cent and not more than the equivalent of 101.0 per cent of 1-hexadecylpyridinium chloride, calculated with reference to the anhydrous substance.

### **CHARACTERS**

A white or almost white powder, slightly soapy to the touch, soluble in water and in alcohol. An aqueous solution froths copiously when shaken.

### **IDENTIFICATION** First identification: B. D.

See the information section on general monographs (cover pages)