

## IDENTIFICATION

A solution containing the equivalent of 100 IU of hyaluronidase in 1 ml of a 9 g/l solution of *sodium chloride R* depolymerises an equal volume of a 10 g/l solution of *sodium hyaluronate BRP* in 1 min at 20 °C as shown by a pronounced decrease in viscosity. This action is destroyed by heating the hyaluronidase at 100 °C for 30 min.

## TESTS

**Appearance of solution.** Dissolve 0.10 g of the substance to be examined in *water R* and dilute to 10 ml. The solution is clear (2.2.1).

**pH (2.2.3).** Dissolve 30 mg in 10 ml of *carbon dioxide-free water R*. The pH of the solution is 4.5 to 7.5.

**Loss on drying (2.2.32).** Not more than 5.0 per cent, determined on 0.500 g by drying at 60 °C at a pressure not exceeding 670 Pa for 2 h.

**Bacterial endotoxins (2.6.14):** less than 0.2 IU per IU of hyaluronidase.

## ASSAY

The activity of hyaluronidase is determined by comparing the rate at which it hydrolyses *sodium hyaluronate BRP* with the rate obtained with the International Standard, or a reference preparation calibrated in International Units, using a slope-ratio assay.

**Substrate solution.** To 0.10 g of *sodium hyaluronate BRP* in a 25 ml conical flask add slowly 20.0 ml of *water R* at 4 °C. The rate of addition must be slow enough to allow the substrate particles to swell (about 5 min). Maintain at 4 °C and stir for at least 12 h. Store at 4 °C and use within 4 days.

**For the test solution and the reference solution, prepare the solution and carry out the dilution at 0 °C to 4 °C.**

**Test solution.** Dissolve a suitable amount of the substance to be examined in *hyaluronidase diluent R* so as to obtain a solution containing  $0.6 \pm 0.3$  IU of hyaluronidase per millilitre.

**Reference solution.** Dissolve a suitable amount of *hyaluronidase BRP* in *hyaluronidase diluent R* so as to obtain a solution containing 0.6 IU of hyaluronidase per millilitre.

In a reaction vessel, mix 1.50 ml of *phosphate buffer solution pH 6.4 R* and 1.0 ml of the substrate solution and equilibrate at  $37 \pm 0.1$  °C. At time  $t_1 = 0$  (first chronometer) add 0.50 ml of the test solution containing  $E_t$  mg of the enzyme to be examined, mix, measure the viscosity of the solution using a suitable viscometer maintained at  $37 \pm 0.1$  °C and record the outflow time  $t_2$  using a second chronometer (graduated in 0.1 second intervals), several times during about 20 min (read on the first chronometer). The following viscometer has been found suitable: Ubbelohde microviscometer (DIN 51 562, Part 2), capillary type MII, viscometer constant about  $0.1 \text{ mm}^2/\text{s}^2$ .

Repeat the procedure using 0.50 ml of the reference solution containing  $E_r$  mg of *hyaluronidase BRP*.

Calculate the viscosity ratio from the expression:

$$\eta_r = \frac{k \times t_2}{0.6915}$$

$k$  = the viscometer constant in  $\text{mm}^2/\text{s}^2$  (indicated on the viscometer);

$t_2$  = the outflow time (in seconds) of the solution;

0.6915 = the kinematic viscosity in  $\text{mm}^2/\text{s}$  of the buffer solution at 37 °C.

Since the enzymatic reaction continues during the outflow time measurements, the real reaction time equals  $t_1 + t_2/2$ , half of the outflow time ( $t_2/2$ ) for which a certain measurement is valid being added to the time  $t_1$  at which the measurement is started. Plot  $(\ln \eta_r)^{-1}$  as a function of the reaction time ( $t_1 + t_2/2$ ) in seconds. A linear relationship is obtained. Calculate the slope for the substance to be examined ( $b_t$ ) and the reference preparation ( $b_r$ ).

Calculate the specific activity in International Units per milligram from the expression:

$$\frac{b_t}{b_r} \times \frac{E_r}{E_t} \times A$$

$A$  = the specific activity of *hyaluronidase BRP* in International Units per milligram.

Carry out the complete procedure at least three times and calculate the average activity of the substance to be examined.

## STORAGE

Store in an airtight container at a temperature of 2 °C to 8 °C. If the substance is sterile, store in a sterile, tamper-proof container.

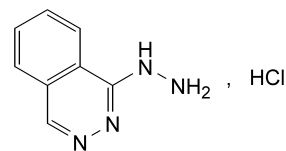
## LABELLING

The label states the activity in International Units per milligram.

01/2008:0829

## HYDRALAZINE HYDROCHLORIDE

## Hydralazini hydrochloridum



$\text{C}_8\text{H}_9\text{ClN}_4$   
[304-20-1]

$M_r$  196.6

## DEFINITION

1-Hydrazinophthalazine hydrochloride.

**Content:** 98.5 per cent to 101.0 per cent (dried substance).

## CHARACTERS

**Appearance:** white or almost white, crystalline powder.

**Solubility:** soluble in water, slightly soluble in ethanol (96 per cent), very slightly soluble in methylene chloride.

mp: about 275 °C, with decomposition.

## IDENTIFICATION

**First identification:** B, E.

**Second identification:** A, C, D, E.

A. Ultraviolet and visible absorption spectrophotometry (2.2.25).

**Test solution.** Dissolve 50 mg in *water R* and dilute to 100 ml with the same solvent. Dilute 2 ml of this solution to 100 ml with *water R*.

**Spectral range:** 220-350 nm.

**Absorption maxima:** at 240 nm, 260 nm, 303 nm and 315 nm.

**Absorbance ratio:**  $A_{240}/A_{303} = 2.0$  to 2.2.

## B. Infrared absorption spectrophotometry (2.2.24).

*Preparation:* discs.

*Comparison:* hydralazine hydrochloride CRS.

C. Dissolve 0.5 g in a mixture of 8 ml of *dilute hydrochloric acid R* and 100 ml of *water R*. Add 2 ml of *sodium nitrite solution R*, allow to stand for 10 min and filter. The precipitate, washed with *water R* and dried at 100-105 °C, melts (2.2.14) at 209 °C to 212 °C.

D. Dissolve about 10 mg in 2 ml of *water R*. Add 2 ml of a 20 g/l solution of *nitrobenzaldehyde R* in *ethanol (96 per cent) R*. An orange precipitate is formed.

E. It gives reaction (a) of chlorides (2.3.1).

## TESTS

**Solution S.** Dissolve 0.5 g in *carbon dioxide-free water R* and dilute to 25 ml with the same solvent.

**Appearance of solution.** The solution is clear (2.2.1) and not more intensely coloured than reference solution GY<sub>6</sub> (2.2.2, Method II).

Dilute 4 ml of solution S to 20 ml with *water R*.

**pH** (2.2.3): 3.5 to 4.2 for solution S.

**Hydrazine.** Thin-layer chromatography (2.2.27).

*Test solution.* Dissolve 0.12 g of the substance to be examined in 4 ml of *water R* and add 4 ml of a 150 g/l solution of *salicylaldehyde R* in *methanol R* and 0.2 ml of *hydrochloric acid R*. Mix and keep at a temperature not exceeding 25 °C for 2-4 h, until the precipitate formed has sedimented. Add 4 ml of *toluene R*, shake vigorously and centrifuge. Transfer the clear supernatant liquid to a 100 ml separating funnel and shake vigorously, each time for 3 min, with 2 quantities, each of 20 ml, of a 200 g/l solution of *sodium metabisulphite R* and with 2 quantities, each of 50 ml, of *water R*. Separate the upper toluene layer which is the test solution.

*Reference solution (a).* Dissolve 12 mg of *hydrazine sulphate R* in *dilute hydrochloric acid R* and dilute to 100.0 ml with the same acid. Dilute 1.0 ml of this solution to 100.0 ml with *dilute hydrochloric acid R*.

*Reference solution (b).* Prepare the solution at the same time and in the same manner as for the test solution, using 1.0 ml of reference solution (a) and 3 ml of *water R*.

*Plate:* TLC silica gel G plate R.

*Mobile phase:* *ethanol (96 per cent) R*, *toluene R* (10:90 V/V).

*Application:* 20 µl of the test solution and reference solution (b).

*Development:* over a path of 10 cm.

*Drying:* in air.

*Detection:* examine in ultraviolet light at 365 nm.

*Limit:*

– *hydrazine:* any yellow fluorescent spot due to hydrazine is not more intense than the corresponding spot in the chromatogram obtained with reference solution (b) (10 ppm).

**Related substances.** Liquid chromatography (2.2.29). *The solutions must be injected within one working day.*

*Test solution.* Dissolve 25.0 mg of the substance to be examined in the mobile phase and dilute to 50.0 ml with the mobile phase.

*Reference solution (a).* Dilute 1.0 ml of the test solution to 100.0 ml with the mobile phase.

*Reference solution (b).* Dilute 10.0 ml of reference solution (a) to 50.0 ml with the mobile phase.

*Reference solution (c).* Dissolve 25.0 mg of *phthalazine R* in the mobile phase and dilute to 50.0 ml with the mobile phase. Dilute 4.0 ml of this solution to 100.0 ml with the mobile phase.

*Reference solution (d).* Dilute a mixture of 4.0 ml of the test solution and 10.0 ml of reference solution (c) to 100.0 ml with the mobile phase.

*Column:*

– *size:*  $l = 0.25$  m,  $\varnothing = 4.6$  mm;

– *stationary phase:* nitrile silica gel for chromatography R1 (10 µm).

*Mobile phase:* mix 22 volumes of *acetonitrile R* and 78 volumes of a solution containing 1.44 g/l of *sodium laurilsulfate R* and 0.75 g/l of *tetrabutylammonium bromide R*, then adjust to pH 3.0 with 0.05 M *sulphuric acid*.

*Flow rate:* 1 ml/min.

*Detection:* spectrophotometer at 230 nm.

*Injection:* 20 µl.

*Run time:* 3 times the retention time of hydralazine.

*Retention time:* hydralazine = about 10 min to 12 min; if necessary, adjust the concentration of acetonitrile in the mobile phase.

*System suitability:*

- the chromatogram obtained with reference solution (d) shows 2 principal peaks;
- *resolution:* minimum 2.5 between the peaks due to hydralazine and phthalazine in the chromatogram obtained with reference solution (d);
- *signal-to-noise ratio:* minimum 3 for the principal peak in the chromatogram obtained with reference solution (b).

*Limit:*

- *any impurity:* for each impurity, not more than the area of the principal peak in the chromatogram obtained with reference solution (b) (0.2 per cent).

**Heavy metals** (2.4.8): maximum 20 ppm.

1.0 g complies with test C. Prepare the reference solution using 2 ml of *lead standard solution (10 ppm Pb) R*.

**Loss on drying** (2.2.32): maximum 0.5 per cent, determined on 1.000 g by drying *in vacuo*.

**Sulphated ash** (2.4.14): maximum 0.1 per cent, determined on 1.0 g.

## ASSAY

Dissolve 80.0 mg in 25 ml of *water R*. Add 35 ml of *hydrochloric acid R* and titrate with 0.05 M *potassium iodate*, determining the end-point potentiometrically (2.2.20), using a calomel reference electrode and a platinum indicator electrode.

1 ml of 0.05 M *potassium iodate* is equivalent to 9.832 mg of C<sub>8</sub>H<sub>9</sub>ClN<sub>4</sub>.

## STORAGE

Protected from light.