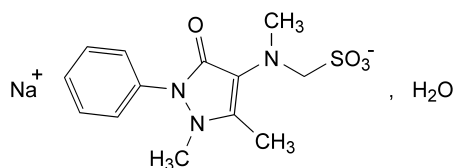


- A. R2 = R3 = R4 = R5 = R6 = H: phenol,  
 B. R2 = CH<sub>3</sub>, R3 = R4 = R5 = R6 = H: 2-methylphenol (*o*-cresol, cresol),  
 C. R2 = R3 = R5 = R6 = H, R4 = CH<sub>3</sub>: 4-methylphenol (*p*-cresol),  
 D. R2 = R6 = CH<sub>3</sub>, R3 = R4 = R5 = H: 2,6-dimethylphenol (2,6-xyleneol),  
 E. R2 = C<sub>2</sub>H<sub>5</sub>, R3 = R4 = R5 = R6 = H: 2-ethylphenol (*o*-ethylphenol),  
 F. R2 = R4 = CH<sub>3</sub>, R3 = R5 = R6 = H: 2,4-dimethylphenol (2,4-xyleneol),  
 G. R2 = R5 = CH<sub>3</sub>, R3 = R4 = R6 = H: 2,5-dimethylphenol (2,5-xyleneol),  
 H. R2 = CH(CH<sub>3</sub>)<sub>2</sub>, R3 = R4 = R5 = R6 = H: 2-(1-methylethyl)phenol,  
 I. R2 = R3 = CH<sub>3</sub>, R4 = R5 = R6 = H: 2,3-dimethylphenol (2,3-xyleneol),  
 J. R2 = R4 = R6 = H, R3 = R5 = CH<sub>3</sub>: 3,5-dimethylphenol (3,5-xyleneol),  
 K. R2 = R3 = R5 = R6 = H, R4 = C<sub>2</sub>H<sub>5</sub>: 4-ethylphenol (*p*-ethylphenol),  
 L. R2 = R5 = R6 = H, R3 = R4 = CH<sub>3</sub>: 3,4-dimethylphenol (3,4-xyleneol),  
 M. R2 = R3 = R5 = CH<sub>3</sub>, R4 = R6 = H: 2,3,5-trimethylphenol.

01/2008:1346

## METAMIZOLE SODIUM

### Metamizolum natricum



C<sub>13</sub>H<sub>16</sub>N<sub>3</sub>NaO<sub>4</sub>S·H<sub>2</sub>O  
 [5907-38-0]

M<sub>r</sub> 351.4

#### DEFINITION

Sodium [(1,5-dimethyl-3-oxo-2-phenyl-2,3-dihydro-1*H*-pyrazol-4-yl)-*N*-methylamino]methanesulphonate monohydrate.

**Content:** 99.0 per cent to 101.0 per cent (dried substance).

#### CHARACTERS

**Appearance:** white or almost white, crystalline powder.

**Solubility:** very soluble in water, soluble in ethanol (96 per cent).

#### IDENTIFICATION

**First identification:** A, D.

**Second identification:** B, C, D.

A. Infrared absorption spectrophotometry (2.2.24).

**Comparison:** metamizole sodium CRS.

B. Dissolve 50 mg in 1 ml of *strong hydrogen peroxide solution R*. A blue colour is produced which fades rapidly and turns to intense red in a few minutes.

C. Place 0.10 g in a test tube, add some glass beads and dissolve the substance in 1.5 ml of *water R*. Add 1.5 ml of *dilute hydrochloric acid R* and place a filter paper wetted with a solution of 20 mg of *potassium iodate R* in 2 ml of *starch solution R* at the open end of the test tube. Heat gently, the evolving vapour of sulphur dioxide colours the filter paper blue. After heating gently for 1 min take a glass rod with a drop of a 10 g/l solution of *chromotropic acid, sodium salt R* in *sulphuric acid R* and place in the opening of the tube. Within 10 min, a blue-violet colour develops in the drop of the reagent.

D. 0.5 ml of solution S (see Tests) gives reaction (a) of sodium (2.3.1).

#### TESTS

**Solution S.** Dissolve 2.0 g in *carbon dioxide-free water R* and dilute to 40 ml with the same solvent.

**Appearance of solution.** Solution S is clear (2.2.1) and immediately after preparation, not more intensely coloured than reference solution BY<sub>6</sub> (2.2.2, *Method I*).

**Acidity or alkalinity.** To 5 ml of solution S, add 0.1 ml of *phenolphthalein solution R1*. The solution is colourless. Not more than 0.1 ml of 0.02 M *sodium hydroxide* is required to change the colour of the indicator to pink.

**Related substances.** Liquid chromatography (2.2.29). *Prepare the solutions immediately before use.*

**Test solution.** Dissolve 50.0 mg of the substance to be examined in *methanol R* and dilute to 10.0 ml with the same solvent.

**Reference solution (a).** Dissolve 10.0 mg of *metamizole impurity A CRS* in *methanol R* and dilute to 20.0 ml with the same solvent.

**Reference solution (b).** Dilute 1.0 ml of reference solution (a) to 20.0 ml with *methanol R*.

**Reference solution (c).** Dissolve 40 mg of *metamizole sodium CRS* in *methanol R* and dilute to 20.0 ml with the same solvent.

**Reference solution (d).** In order to prepare impurity C *in situ*, boil 10 ml of reference solution (c) under reflux for 10 min. Allow to cool to room temperature and dilute to 20.0 ml with *methanol R*.

**Reference solution (e).** To 6 ml of reference solution (a) add 1 ml of reference solution (c).

#### Column:

– *size:* *l* = 0.25 m, Ø = 4.6 mm;

– *stationary phase:* base-deactivated octadecylsilyl silica gel for chromatography R (5 µm).

**Mobile phase:** mix 28 volumes of *methanol R* and 72 volumes of a buffer solution prepared as follows: mix 1000 volumes of a 6.0 g/l solution of *sodium dihydrogen phosphate R* and 1 volume of *triethylamine R*, then adjust to pH 7.0 with *strong sodium hydroxide solution R*.

**Flow rate:** 1.0 ml/min.

**Detection:** spectrophotometer at 254 nm.

**Injection:** 10 µl of the test solution and reference solutions (b), (d) and (e).

**Run time:** 3.5 times the retention time of metamizole.

**Elution order:** impurity A, metamizole, impurity B, impurity C, impurity D.

**System suitability:** reference solution (e):

- **resolution:** minimum 2.5 between the peaks due to impurity A and metamizole.

**Limits:**

- **impurity C:** not more than the area of the principal peak in the chromatogram obtained with reference solution (b) (0.5 per cent);
- **impurities A, B, D:** for each impurity, not more 0.4 times the area of the principal peak in the chromatogram obtained with reference solution (b) (0.2 per cent);
- **total:** not more than the area of the principal peak in the chromatogram obtained with reference solution (b) (0.5 per cent);
- **disregard limit:** 0.05 times the area of the principal peak in the chromatogram obtained with the reference solution (b) (0.025 per cent).

**Sulphates (2.4.13):** maximum 0.1 per cent.

Dissolve 0.150 g in *distilled water R* and dilute to 15 ml with the same solvent.

**Heavy metals (2.4.8):** maximum 20 ppm.

Dissolve 2.0 g in *water R* and dilute to 20 ml with the same solvent. 12 ml of the freshly prepared solution complies with test A. Prepare the reference solution using *lead standard solution (2 ppm Pb) R*.

**Loss on drying (2.2.32):** 4.9 per cent to 5.3 per cent, determined on 1.000 g by drying in an oven at 105 °C.

#### ASSAY

Dissolve 0.200 g in 10 ml of 0.01 M hydrochloric acid previously cooled in iced water and titrate immediately, dropwise, with 0.05 M iodine. Before each addition of 0.05 M iodine dissolve the precipitate by swirling. At the end of the titration add 2 ml of *starch solution R* and titrate until the blue colour of the solution persists for at least 2 min. The temperature of the solution during the titration must not exceed 10 °C.

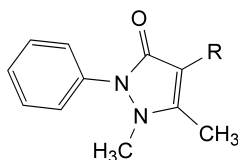
1 ml of 0.05 M iodine is equivalent to 16.67 mg of C<sub>13</sub>H<sub>16</sub>N<sub>3</sub>NaO<sub>4</sub>S.

#### STORAGE

Protected from light.

#### IMPURITIES

**Specified impurities:** A, B, C, D.



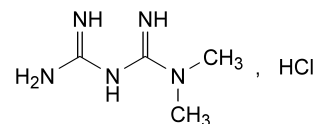
- A. R = NHCHO: 4-formylamino-1,5-dimethyl-2-phenyl-1,2-dihydro-3H-pyrazol-3-one,  
 B. R = NH<sub>2</sub>: 4-amino-1,5-dimethyl-2-phenyl-1,2-dihydro-3H-pyrazol-3-one,  
 C. R = NHCH<sub>3</sub>: 4-methylamino-1,5-dimethyl-2-phenyl-1,2-dihydro-3H-pyrazol-3-one,

D. R = N(CH<sub>3</sub>)<sub>2</sub>: 4-dimethylamino-1,5-dimethyl-2-phenyl-1,2-dihydro-3H-pyrazol-3-one.

01/2008:0931  
corrected 6.0

## METFORMIN HYDROCHLORIDE

### Metformini hydrochloridum



C<sub>4</sub>H<sub>12</sub>ClN<sub>5</sub>  
[1115-70-4]

M<sub>r</sub> 165.6

#### DEFINITION

1,1-Dimethylbiguanide hydrochloride.

**Content:** 98.5 per cent to 101.0 per cent (dried substance).

#### CHARACTERS

**Appearance:** white or almost white crystals.

**Solubility:** freely soluble in water, slightly soluble in alcohol, practically insoluble in acetone and in methylene chloride.

#### IDENTIFICATION

**First identification:** B, E.

**Second identification:** A, C, D, E.

A. Melting point (2.2.14): 222 °C to 226 °C.

B. Infrared absorption spectrophotometry (2.2.24).

**Preparation:** discs of *potassium chloride R*.

**Comparison:** *metformin hydrochloride CRS*.

C. Thin-layer chromatography (2.2.27).

**Test solution.** Dissolve 20 mg of the substance to be examined in *water R* and dilute to 5 ml with the same solvent.

**Reference solution.** Dissolve 20 mg of *metformin hydrochloride CRS* in *water R* and dilute to 5 ml with the same solvent.

**Plate:** *TLC silica gel G plate R*.

**Mobile phase:** upper layer of a mixture of 10 volumes of *glacial acetic acid R*, 40 volumes of *butanol R* and 50 volumes of *water R*.

**Application:** 5 µl.

**Development:** over a path of 15 cm.

**Drying:** at 100-105 °C for 15 min.

**Detection:** spray with a mixture of equal volumes of a 100 g/l solution of *sodium nitroprusside R*, a 100 g/l solution of *potassium ferricyanide R* and a 100 g/l solution of *sodium hydroxide R*, prepared 20 min before use.

**Results:** the principal spot in the chromatogram obtained with the test solution is similar in position, colour and size to the principal spot in the chromatogram obtained with the reference solution.

- D. Dissolve about 5 mg in *water R* and dilute to 100 ml with the same solvent. To 2 ml of the solution add 0.25 ml of *strong sodium hydroxide solution R* and 0.10 ml of *α-naphthol solution R*. Mix and allow to stand in iced water for 15 min. Add 0.5 ml of *sodium hypobromite solution R* and mix. A pink colour develops.  
 E. It gives reaction (a) of chlorides (2.3.1).