- C. R = R2 = H, R4 = O-CH $_2$ -CH $_2$ -N(CH $_3$) $_2$: 2-[4-[(*EZ*)-1,2-diphenylethenyl]phenoxy]-*N*,*N*-dimethylethanamine,
- D. $R = CH_3$, R2 = H, $R4 = O-CH_2-CH_2-N(CH_3)_2$: 2-[4-[(EZ)-1,2-diphenylprop-1-enyl]phenoxy]-N,N-dimethylethanamine,
- E. $R = C_2H_5$, $R2 = O-CH_2-CH_2-N(CH_3)_2$, R4 = H: 2-[2-[(EZ)-1,2-diphenylbut-1-enyl]phenoxy]-N,N-dimethylethanamine,

F. 2-[4-[(*Z*)-1,2-diphenylbut-1-enyl]phenoxy]-*N*-methylethanamine,

H. 2-[4-[(*Z*)-1-[4-[(*RS*)-[4-[2-(dimethylamino)ethoxy]-phenyl]phenylmethyl]phenyl]-2-phenylbut-1-enyl]-phenoxy]-*N*,*N*-dimethylethanamine.

01/2008:2131

TAMSULOSIN HYDROCHLORIDE

Tamsulosini hydrochloridum

 $C_{20}H_{29}CIN_2O_5S$

 $M_{r}445.0$

DEFINITION

5-[(2R)-2-[[2-(2-Ethoxyphenoxy)ethyl]amino] propyl]-2-methoxybenzenesulfonamide hydrochloride.

Content: 98.5 per cent to 101.0 per cent (dried substance).

CHARACTERS

Appearance: white or almost white powder.

Solubility: slightly soluble in water, freely soluble in formic acid, slightly soluble in anhydrous ethanol.

mp: about 230 °C.

IDENTIFICATION

Carry out either tests A, C, D or tests A, B, D.

- A. Infrared absorption spectrophotometry (2.2.24). Comparison: tamsulosin hydrochloride CRS.
- B. Specific optical rotation (2.2.7): -17.5 to -20.5 (dried substance).

Dissolve with heating 0.15~g in *water R* and dilute to 20.0~ml with the same solvent.

- C. Enantiomeric purity (see Tests).
- D. Dissolve with heating 0.75 g in *water R* and dilute to 100.0 ml with the same solvent. Take 5 ml of the solution and cool in an ice-bath. Add 3 ml of *dilute nitric acid R* and shake. Allow to stand at room temperature for 30 min and filter. The filtrate gives reaction (a) of chlorides (2.3.1).

TESTS

Related substances.

A. Impurities eluting before tamsulosin. Liquid chromatography (2.2.29).

Test solution. Dissolve 50.0 mg of the substance to be examined in the mobile phase and dilute to 10.0 ml with the mobile phase.

Reference solution (a). Dilute 1.0 ml of the test solution to 100.0 ml with the mobile phase. Dilute 1.0 ml of this solution to 10.0 ml with the mobile phase.

Reference solution (b). Dissolve 4 mg of tamsulosin impurity D CRS and 4 mg of the substance to be examined in the mobile phase and dilute to 20.0 ml with the mobile phase. Dilute 2.0 ml of this solution to 20.0 ml with the mobile phase.

Reference solution (c). Dissolve 4 mg of tamsulosin impurity H CRS and 4 mg of the substance to be examined in the mobile phase and dilute to 20.0 ml with the mobile phase. Dilute 2.0 ml of this solution to 20.0 ml with the mobile phase.

Column:

- size: l = 0.15 m, $\emptyset = 4.6$ mm;
- stationary phase: octadecylsilyl silica gel for chromatography R (5 µm);
- temperature: 40 °C.

Mobile phase: dissolve 3.0 g of sodium hydroxide R in a mixture of 8.7 ml of perchloric acid R and 1.9 litres of water R; adjust to pH 2.0 with 0.5 M sodium hydroxide and dilute to 2 litres with water R; to 1.4 litres of this solution, add 600 ml of acetonitrile R.

Flow rate: 1.3 ml/min.

Detection: spectrophotometer at 225 nm.

Injection: 10 µl of the test solution and reference solutions (a) and (b).

Run time: 1.5 times the retention time of tamsulosin (retention time = about 6 min).

System suitability: reference solution (b):

 resolution: minimum 6 between the peaks due to impurity D and tamsulosin.

Limits:

- unspecified impurities: for each impurity, not more than the area of the principal peak in the chromatogram obtained with reference solution (a) (0.10 per cent);
- disregard limit: 0.5 times the area of the principal peak in the chromatogram obtained with reference solution (a) (0.05 per cent).
- B. Impurities eluting after tamsulosin. Liquid chromatography (2.2.29) as described in test A with the following modifications.

Mobile phase: dissolve 3.0 g of sodium hydroxide R in a mixture of 8.7 ml of perchloric acid R and 1.9 litres of water R; adjust to pH 2.0 with 0.5 M sodium hydroxide and dilute to 2 litres with water R; add 2 litres of acetonitrile R.

Flow rate: 1.0 ml/min.

Injection: 10 µl of the test solution and reference solutions (a) and (c).

Run time: 5 times the retention time of tamsulosin (retention time = about 2.5 min).

System suitability: reference solution (c):

 resolution: minimum 2 between the peaks due to tamsulosin and impurity H.

Limits

- unspecified impurities: for each impurity, not more than the area of the principal peak in the chromatogram obtained with reference solution (a) (0.10 per cent);
- sum of impurities eluting before tamsulosin in test A and after tamsulosin in test B: not more than twice the area of the principal peak in the chromatogram obtained with reference solution (a) (0.2 per cent);
- disregard limit: 0.5 times the area of the principal peak in the chromatogram obtained with reference solution (a) (0.05 per cent).

Enantiomeric purity. Liquid chromatography (2.2.29).

Test solution. Dissolve 50.0 mg of the substance to be examined in *methanol R* and dilute to 25.0 ml with the same solvent.

Reference solution (a). Dilute 1.0 ml of the test solution to 100.0 ml with *methanol R*. Dilute 1.0 ml of this solution to 10.0 ml with *methanol R*.

Reference solution (b). Dissolve 5.0 mg of tamsulosin racemate CRS in methanol R and dilute to 25.0 ml with the same solvent. Dilute 2.0 ml of this solution to 10.0 ml with methanol R.

Column:

- size: l = 0.25 m, $\emptyset = 4.6$ mm;

- stationary phase: silica gel OD for chiral separations R;

temperature: 40 °C.

Mobile phase: diethylamine R, methanol R, anhydrous ethanol R, hexane R (1:150:200:650 V/V/V/V).

Flow rate: 0.5 ml/min.

Detection: spectrophotometer at 225 nm.

Injection: 10 µl.

Relative retention with reference to tamsulosin (retention time = about 14 min): impurity G = about 0.8.

System suitability: reference solution (b):

 resolution: minimum 2 between the peaks due to impurity G and tamsulosin.

Limit

 impurity G: not more than the area of the principal peak in the chromatogram obtained with reference solution (a) (0.1 per cent).

Heavy metals (2.4.8): maximum 20 ppm.

1.0 g complies with test C. Prepare the reference solution using 2 ml of *lead standard solution (10 ppm Pb) R*.

Loss on drying (2.2.32): maximum 0.5 per cent, determined on 1.000 g by drying in an oven at 105 °C for 2 h.

Sulphated ash (2.4.14): maximum 0.1 per cent, determined on 1.0 g.

ASSAY

Dissolve 0.350 g in 5.0 ml of anhydrous formic acid R, add 75 ml of a mixture of 2 volumes of acetic anhydride R and 3 volumes of glacial acetic acid R. Titrate immediately with 0.1 M perchloric acid, determining the end-point potentiometrically (2.2.20). Carry out a blank titration. 1 ml of 0.1 M perchloric acid is equivalent to 44.50 mg of $C_{20}H_{29}ClN_2O_5S$.

IMPURITIES

Specified impurities: G.

Other detectable impurities (the following substances would, if present at a sufficient level, be detected by one or other of the tests in the monograph. They are limited by the general acceptance criterion for other/unspecified impurities and/or by the general monograph Substances for pharmaceutical use (2034). It is therefore not necessary to identify these impurities for demonstration of compliance. See also 5.10. Control of impurities in substances for pharmaceutical use): A, B, C, D, E, F, H, I.

A. 5-[(2R)-2-[bis[2-(2-ethoxyphenoxy)ethyl]amino]propyl]-2-methoxybenzenesulfonamide,

B. 5-[(2R)-2-aminopropyl]-2-methoxybenzenesulfonamide,

- C. R1 = SO_2 -NH₂, R2 = H: 2-methoxy-5-[(2R)-2-[(2-phenoxyethyl)amino]propyl]benzenesulfonamide,
- D. $R1 = SO_2$ - NH_2 , $R2 = OCH_3$: 2-methoxy-5-[(2R)-2-[[2-(2-methoxyphenoxy)ethyl]amino]propyl]benzenesulfonamide,
- H. R1 = H, R2 = OC_2H_5 : (2*R*)-*N*-[2-(2-ethoxyphenoxy)ethyl]-1-(4-methoxyphenyl)propan-2-amine,

$$\begin{array}{c|c} \text{O} & \text{O} \\ \hline & \text{S} & \text{NH}_2 \\ \\ \text{OCH}_3 \end{array}$$

E. 5-formyl-2-methoxybenzenesulfonamide,

F. $R = NH_2$: 2-(2-ethoxyphenoxy)ethanamine,

I. R = Br: 1-(2-bromoethoxy)-2-ethoxybenzene,

G. 5-[(2*S*)-2-[[2-(2-ethoxyphenoxy)ethyl]amino]propyl]-2-methoxybenzenesulfonamide.

01/2008:1477 corrected 6.0

TANNIC ACID

Tanninum

DEFINITION

Mixture of esters of glucose with gallic acid and 3-galloylgallic acid.

CHARACTERS

Appearance: yellowish-white or slightly brown amorphous light powder or shiny plates.

Solubility: very soluble in water, freely soluble in acetone, in ethanol (96 per cent) and in glycerol (85 per cent), practically insoluble in methylene chloride.

IDENTIFICATION

- A. Dilute 0.1 ml of solution S (see Tests) to 5 ml with water R. Add 0.1 ml of ferric chloride solution R1. A blackish-blue colour is produced which becomes green on the addition of 1 ml of dilute sulphuric acid R.
- B. To 1 ml of solution S, add 3 ml of a 1 g/l solution of *gelatin R*. The mixture becomes turbid and a flocculent precipitate is formed.
- C. Dilute 0.1 ml of solution S to 5 ml with *water R*. Add 0.3 ml of *barium hydroxide solution R*. A greenish-blue precipitate is formed.

TESTS

Solution S. Dissolve 4.0 g in *carbon dioxide-free water R* and dilute to 20 ml with the same solvent.

Appearance of solution. Solution S is not more opalescent than reference suspension II (*2.2.1*).

Dextrins, gum, salts, sugars. To 2 ml of solution S, add 2 ml of *ethanol (96 per cent) R*. The solution is clear. Add 1 ml of *ether R*. The solution remains clear for at least 10 min.

Resins. To 5 ml of solution S, add 5 ml of *water R*. The mixture remains clear (2.2.1) for at least 15 min.

Loss on drying (2.2.32): maximum 12.0 per cent, determined on 0.200 g by drying at 105 °C.

Sulphated ash (2.4.14): maximum 0.1 per cent, determined on 1.0 g.

STORAGE

Protected from light.

01/2008:0460 corrected 6.0

TARTARIC ACID

Acidum tartaricum

$$HO_2C$$
 CO_2H HOH

C₄H₆O₆ [87-69-4]

 $M_{\rm r}$ 150.1

DEFINITION

(2R,3R)-2,3-Dihydroxybutanedioic acid.

Content: 99.5 per cent to 101.0 per cent (dried substance).

CHARACTERS

Appearance: white or almost white, crystalline powder or colourless crystals.

Solubility: very soluble in water, freely soluble in ethanol (96 per cent).

IDENTIFICATION

A. Solution S (see Tests) is strongly acid (2.2.4).

B. It gives the reactions of tartrates (2.3.1).

TESTS

Solution S. Dissolve 5.0 g in *distilled water R* and dilute to 50 ml with the same solvent.

Appearance of solution. Solution S is clear (2.2.1) and not more intensely coloured than reference solution Y_6 (2.2.2, *Method II*).

Specific optical rotation (2.2.7): + 12.0 to + 12.8 (dried substance).

Dissolve 5.00 g in *water R* and dilute to 25.0 ml with the same solvent.

Oxalic acid: maximum 350 ppm, calculated as anhydrous oxalic acid.

Dissolve 0.80 g in 4 ml of *water R*. Add 3 ml of *hydrochloric acid R* and 1 g of *zinc R* in granules and boil for 1 min. Allow to stand for 2 min. Collect the liquid in a test-tube containing 0.25 ml of a 10 g/l solution of *phenylhydrazine hydrochloride R* and heat to boiling. Cool rapidly, transfer to a graduated cylinder and add an equal volume of *hydrochloric acid R* and 0.25 ml of a 50 g/l solution of *potassium ferricyanide R*. Shake and allow to stand for 30 min. Any pink colour in the solution is not more intense than that in a standard prepared at the same time in the same manner using 4 ml of a 0.1 g/l solution of *oxalic acid R*.

Chlorides (2.4.4): maximum 100 ppm.

Dilute 5 ml of solution S to 15 ml with water R.